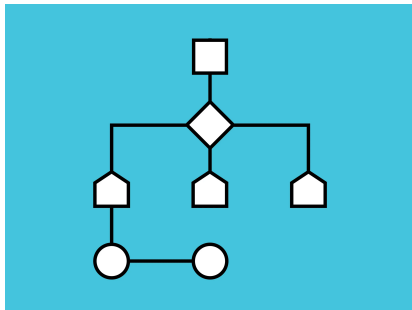


How to derive the Gillespie algorithm



From the master equation / Kolmogorov forward equation ?

- ▶ $p_{ij}(t) = P(X(t) = j \mid X(0) = i)$ transition probabilities.
- ▶ $P(t) = (p_{ij}(t))$
- ▶ $q_{ij} = \lim_{\Delta t \rightarrow 0} \frac{p_{ij}(\Delta t) - p_{ij}(0)}{\Delta t}$ transition rates.
- ▶ $Q = (q_{ij})$

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No! From a more fundamental equation: the **fundamental premise of stochastic kinetics**:

$a_i dt$ = the probability that the reaction i will occur in the infinitesimal time interval $(t, t + dt)$

where $a_i(\mathbf{x})$ is the total rate (or the propensity) of the reaction i

Most straightforward method.

1. Discretize the time-axis in small enough time-steps Δt .
2. For each Δt , check for each reaction whether they occur or not.

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- 1. When does the next reaction occur?**
- 2. Which reaction occurs next?**

Joseph L. Doob (1942 and 1945)



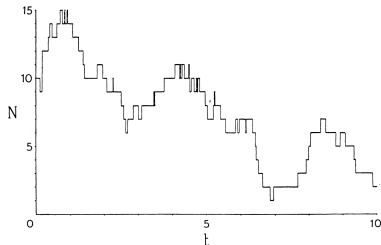
AN ARTIFICIAL REALIZATION OF A SIMPLE "BIRTH-AND-DEATH" PROCESS

By DAVID G. KENDALL

Magdalen College, Oxford

[Received November 1st, 1949]

Manchester Mark 1



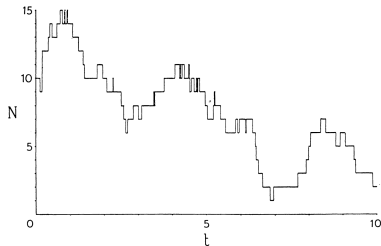
The graph shows the variation with time (t) in the number of individuals (N) in an artificial population developing in accordance with the simple birth-and-death process ($\lambda = \mu = \frac{1}{2}$).

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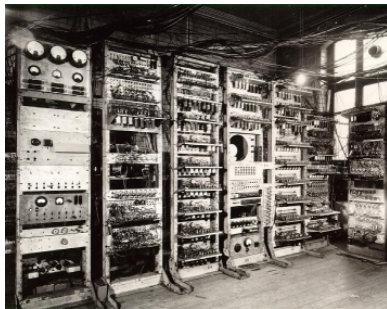
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Manchester Mark 1



JOURNAL OF COMPUTATIONAL PHYSICS **22**, 403–434 (1976)

**A General Method for Numerically Simulating
the Stochastic Time Evolution
of Coupled Chemical Reactions**

DANIEL T. GILLESPIE

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Received January 27, 1976; revised April 21, 1976

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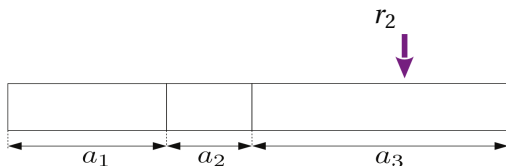
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- ▶ draw a reaction i in a discrete distribution for which $P(X = i) = \frac{a_i}{a_0}$.



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6. Save (\mathbf{x}, t) as desired and return to Step 1, or else end the simulation

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One can prove that the joint probability $p(\tau, i)$ which is generated by this method is the one we want:

$$\begin{aligned} p(\tau, i) &= P(\tau < \tau_i < \tau + dt) \times P(\tau_j > \tau \quad \forall j \neq i) \\ &= e^{-a_0 \tau} \times a_i \end{aligned}$$

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⚠ The first reaction might change the propensities a_i of other reactions making previous calculations obsolete.